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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: a. copper(II) carbonate b. sodium sulfite c. ammonium phosphate d. tin(IV) sulfide e. nitrous acid 2. Write the Stock system names for the following compounds: a. $\text{Mg}(\text{ClO}_4)_2$ b. $\text{Fe}(\text{NO}_3)_2$

7 Chemical Formulas and Chemical Compounds

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54 CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS MODERN CHEMISTRY ... CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a.

7 Chemical Formulas and Chemical Compounds

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Chemical Names & Formulas, Oxidation Numbers, Using Chemical Formulas, & Determining Chemical Formulas Terms. Terms in this set (32) monatomic ion. an ion formed from a single atom. (remember to change nonmetals to -ide ending)

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MODERN CHEMISTRY CHAPTER 2 MIXED REVIEW 15 HRW

material copyrighted under notice appearing earlier in this work. Name Date Class . MIXED REVIEW continued 6. Three students were asked to determine the volume of a liquid by a method of their choosing. Each did three trials. The table below shows the results.

2 Measurements and Calculations

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An easier way to understand reactions is to once again consider cooking, but this time, think about making a pizza Modern chemistry chapter 13 mixed review answers. The foundation of pizza is, of course, the crust, and pizza dough is the raw . . .

Modern Chemistry Chapter 13 Mixed Review Answers

CHAPTER 14 REVIEW . Acids and Bases. SHORT ANSWER Answer the following questions in the space provided. 1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1: $\text{H}_2\text{SO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{HSO}_3^-(\text{aq})$ b. Which stage of ionization usually produces more ions? Explain your answer.

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CHAPTER 14 REVIEW Acids and Bases

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Chemistry Principles of Modern Chemistry Lemon juice contains citric acid ($C_6H_8O_7$) . What species serves as a base when lemon juice is mixed with baking soda (sodium hydrogen carbonate) during the preparation of some lemon cookies?

Lemon juice contains citric acid ($C_6H_8O_7$) . What ...

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A 1.0-L solution that is 4.2 M in ammonia is mixed with 26.7 g of ammonium chloride.. a What is the hydroxide-ion concentration of this solution? b 0.075 mol of MgCl_2 is added to the above solution. Assume that there is no volume change.

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