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QUESTIONS:

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Questions & Answers 1.
Which set of diagrams
shows a balanced
chemical equation? A.
Diagram A B. Diagram
B C. Diagram C D.
Diagram D

Explanation: In answer
C, each side contains 4
atoms of hydrogen (H),
2 atoms of ...

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a boon for fast paced
doctors for the reason
that they free the
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earning appointments,
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answering a variety of
other routine issues
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Practice balancing
chemical equations by
changing the
coefficients of
reactants and
products. As the
equation is
manipulated, the
amount of each
element is shown as

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individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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have become a boon

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for fast paced doctors
for the reason that

they free the doctors
from multiple routine
jobs like earning
appointments, giving
directions towards
clinic and answering a
variety of other routine
issues from patients.

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balancing chemical
equations by changing

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the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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the type of thesis
template one has
chosen, particularly if it
contained highly
detailed writing
instructions, the

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completed thesis may contain odd transitions or sentences that the template seemed to require.

Balancing Chemical Equations Gizmo Answers

To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in "H₂+O₂" in the Reactants box and

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“H₂O” in the Products box. This...

Student Exploration- Balancing Chemical Equations (ANSWER

...

The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and

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water molecules until
the equation is
balanced. When you
are done, turn on

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...

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look

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at the floating molecules below the initial reaction: $H_2 + O_2 \Rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? H: 2 O: 1 4. In general, what does a subscript (such as the "2" in H

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The chemical equation has the products on the right side, while the reactants are written on the left side. Both of them are separated by an arrow. For instance, $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ denotes that there are four atoms of hydrogen and 2 atoms of oxygen on both sides of the equation. The amount

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of reactants must be equal to the amount of products.

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This is the fourth

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lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to

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describe and predict
chemical reactions.

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